

# Chapter 1.7: Transport properties

- The properties so far considered were for **thermal** changes of steam;
- **Transport** properties are of use for working out the flow behaviour;
- These are:
  - density,
  - Viscosity,
  - Prandtl number (to be met in convective heat transfer,
  - thermal conductivity;
- They set the heat and mass transfer characteristics;
- There are other electrical properties and optical properties, but they are not of relevance to this course.

# Density

- Density is affected by temperature and pressure
- We know perfect gases have the equation of state
- In the case of steam, the steam tables tell us the density because it is not a perfect gas
- Likewise for refrigerant R134a
- See the video for the dense gas carbon hexafluoride

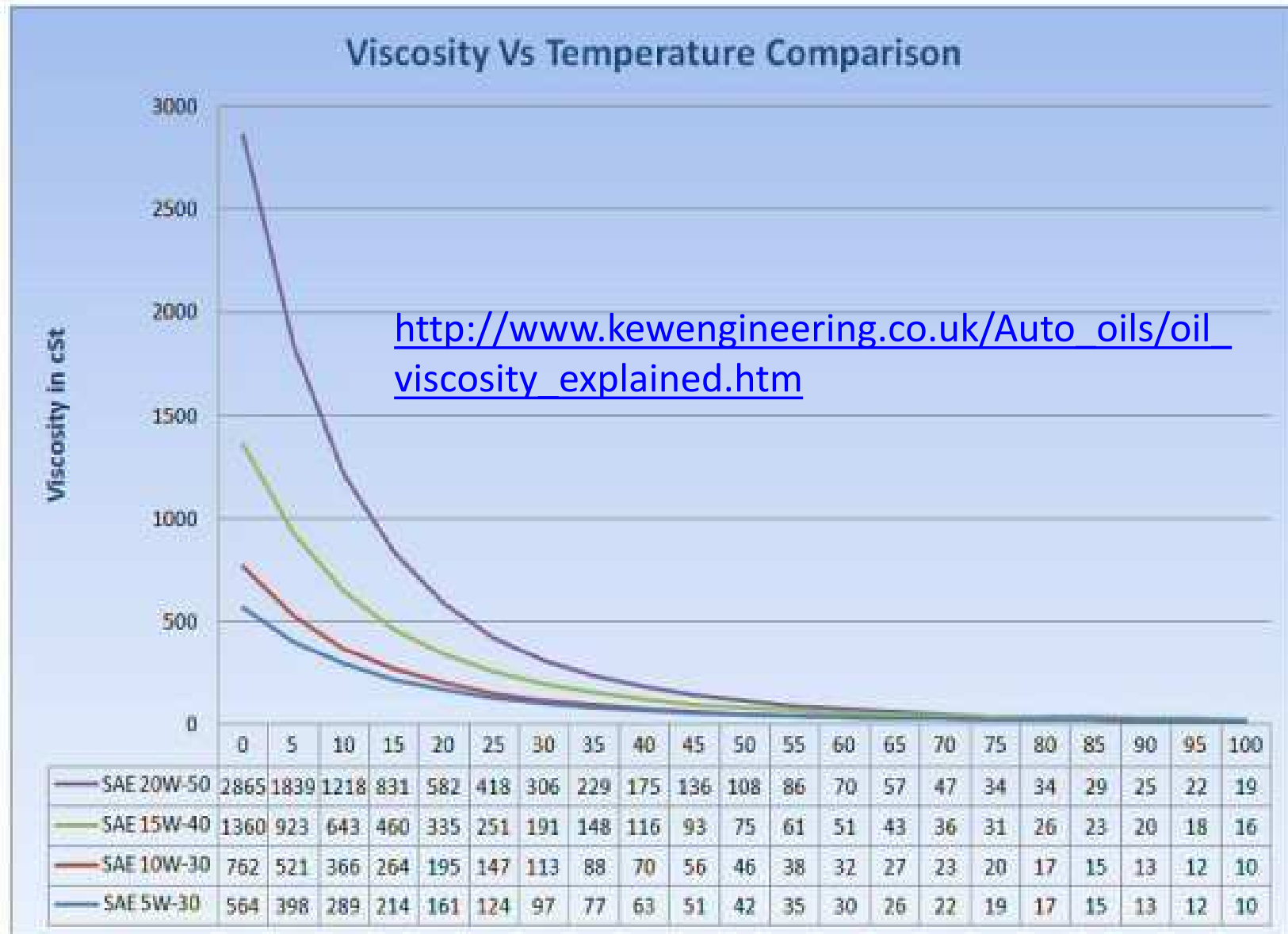
<https://www.youtube.com/watch?v=DzLX96VWTkc&feature=youtu.be>

# Viscosity

- Viscosity in all fluids varies with temperature
- On page 10 of the tables there is viscosity of steam and water liquid stated in dynamic form (i.e. as a mass related figure),  $\mu$  [ $\text{kg}\cdot\text{m}^{-1}\text{s}^{-1}$ ]
- The specific volume [ $\text{m}^3\cdot\text{kg}^{-1}$ ] also on p.10 for steam can be used to invert and get density
- Density can be divided into dynamic viscosity to get kinematic viscosity (i.e. a measure of diffusiveness of the fluid),  $\nu$  [ $\text{m}^2\cdot\text{s}^{-1}$ ]

# Engine oil, viscosity

- An interesting case is engine oil (not in our tables), from <https://www.youtube.com/watch?v=V5a4kP-5Jiw>



# Thermal conductivity, $k$ [W/mK]

- Just as there are solids which have varying levels of thermal conductivity [ $\text{W}\cdot\text{m}^{-1}\text{K}^{-1}$ ], liquids and gases also conduct via molecular vibrations
- The units show it is the energy transferred per m thickness of the material per K temperature difference over the thickness
- The thermal conduction is often slower than convection due to carrying hot fluid to areas of cold
- But in relatively motionless fluids, conduction is important, and especially at the boundary between solid and liquid where the velocity is zero, it entirely determines how much heat transfer can occur into the fluid from the wall <https://www.youtube.com/watch?v=UBcZsWJ5TNA&feature=youtu.be>

# What you can know from tables

- Tables p.1-4 are **saturated steam**
  - Boiling point temperatures vs. pressures
  - Specific volume of vapour
  - Enthalpy and entropy
- Tables p.5-9 are **superheated steam**
  - Enthalpy and entropy most useful
- Tables p.10 are **water/steam**
  - Specific volume, specific heat, viscosity, thermal conductivity, Prandtl number
- Tables p.15 is R134a **refrigerant**
  - Saturation temperature/pressure, enthalpy, entropy, specific volume
- Tables p.24 is the **international standard atmosphere**